# All That Matters Exciting developments in the physical sciences

Enter search keyword

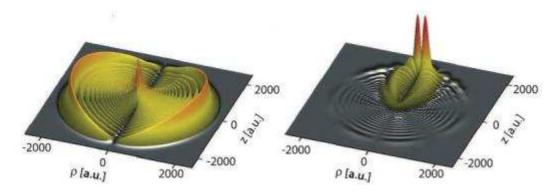
- <u>Home</u>
- <u>About</u>
- Contact me
- <u>Disclaimer</u>



## **Electrons out of balance**

November 27, 2011

Atomic Physics, Condensed Matter Physics



Permanent electric polarization. Left: the electron density of the rubidium Rydberg electron in one of the atoms of the molecule. The asymmetry is hardly visible. Right: The same electron density, but with the density of the Rydberg atom on its own subtracted. The difference clearly shows an asymmetric distribution of electrons roughly at the position of the second rubidium atom, causing an electric polarization. (c) 2011 Science Magazine

Apply an electric field to a material, and its positive and negative charges will separate, creating an electric polarization. This is the fundamental effect behind capacitors used in electronics as well as in ferroelectrics used in some computer memories. In the latter case, to achieve a permanent electric polarization, the positive and negative charges need to be shifted permanently. This is the case if in a crystal all positive and all negative ions in a crystal are shifted in the same way with respect to each other.

Electrons out of balance | All That Matters

+ + +- - -+ + +- - -+ + +

The separation of positive and negative ions in a crystal can lead to a permanent electric polarization. A similar effect has now been achieved with electrons.

In a paper in *Science* from last week, researchers from the group of <u>Tilman Pfau</u> at the University of Stuttgart in Germany with colleagues from a number of other institutions have now demonstrated an entirely new way of achieving permanent electric polarization – namely by using electrons and not ions. This effect is remarkable, because electrons usually are much more mobile than atoms. Normally, any charge imbalance in the electron distribution of a material or molecule is easily neutralized simply by shifting electrons in the molecule around. At the same time, looking far ahead, such electron-based effects could lead to applications where the electric polarization needs to switch ultrafast.

However, the molecule studied by the researchers is quite different to usual molecules. It is formed by two rubidium atoms, which means that normally it should not show any electric polarization, simply because both atoms in the molecule are identical and for symmetry reasons no positive or negative ions would form in the first place.

But although they are both rubidium atoms, here there is a crucial difference in the electronic states. One of the atoms is in its energetic ground state, while the other is a so-called Rydberg atom, which means that its outermost electron is excited into a very high energy state and circulates the atom's nucleus at a large distance. Rydberg atoms are huge in comparison. Here, the rubidium atom is roughly about 50 nanometres in size, corresponding to about 1,000 times the size of a oxygen molecule - and is larger also than the transistors in modern computer chips.

Obviously, for molecules made of an atom in its ground state and one in one of these Rydberg states there is a large imbalance in the electronic states. But still, how is it possible that in such a molecule any electric polarization forms? With one electron being that far out, the interaction between the electronic state of the two atoms is extremely small. The reason a polarization forms in the molecule is that the electronic state of the Rydberg atom extends so far out that it extends across the other rubidium atom. The presence of this atom alters the Rydberg atom's electronic states at the location of the regular atom. That way, the symmetry of the electronic states of the molecule is broken, and this imbalance is the origin of the permanent electric polarization. To detect such a small imbalance, the researchers measured the shift of the molecule's

optical resonances in an electric field, which clearly confirms the effect.

More generally, electron-based polarizations could be very interesting because the electrons of a molecule are much more mobile than atoms themselves, so that the switching of the polarization – for example turning it around by 180 degrees – should occur much faster than in conventional materials, where the polarization is based on ions. This could be interesting for applications such as ultrafast data storage, where up and down of the polarization would represent the 1 and 0 of a computer bit.

Of course, we are still far away from such dreams. The Rydberg atoms are stable only at ultralow temperatures a few millionth of a degree above absolute zero. And also, here this effect seems again down to atomic positions, even if somewhat indirectly; it is the position of the second atom that determines the electric polarization. On the other hand, there are theoretical predictions where some molecules could also show electron-based electric polarization effects at much higher temperatures. Therefore, it seems to me this study is only the beginning for electron-based polarization effects, and we should look forward to similar observations being made in other molecules, too.

Reference:

Li, W., Pohl, T., Rost, J., Rittenhouse, S., Sadeghpour, H., Nipper, J., Butscher, B., Balewski, J., Bendkowsky, V., Low, R., & Pfau, T. (2011). A Homonuclear Molecule with a Permanent Electric Dipole Moment *Science*, *334* (6059), 1110-1114 DOI: <u>10.1126/science.1211255</u>

Share this:				

Like this:

Be the first to like this.

 $\frac{\text{ferroelectricity, ferromagnet, Rydberg atoms}}{\leftarrow \text{Science doesn't shoot from the hip}}$ Whither organic solar cells?  $\rightarrow$ 

#### **Trackbacks/Pingbacks**

1. Electrons out of balance | Science News | Scoop.it - November 28, 2011

[...] Electrons out of balance Apply an electric field to a material, and its positive and negative charges will separate, creating an electric polarization. This is the fundamental effect behind capacitors used in electronics a... Source: blog.joerg.heber.name [...]

### Leave a Comment

Enter your comment here...



Welcome to my personal blog, where I write about the latest exciting developments in the physical sciences. As Managing Editor at Nature Communications I bring my own perspective to the latest scientific trends.

Connect with me

Subscribe to my RSS feed and social profiles to receive updates.



#### Email subscription

Enter your email address to follow this blog and receive notifications of new posts by email.

#### Subscribe

- <u>Latest</u>
- <u>Popular</u>
- <u>Comments</u>
- <u>Tags</u>
- The curse of the obsession with doctorate degrees February 5, 2013
- Lithium-ion batteries and the Boeing 787 Dreamliner January 19, 2013
- Measuring mass by telling the time January 13, 2013
- Atoms at negative absolute temperatures January 4, 2013
- <u>Samarium hexaboride</u> December 12, 2012

#### Archive

Select Month

#### Categories

- Atomic Physics (8)
- <u>Biomaterials</u> (3)
- <u>Books</u> (5)
- <u>Chemistry</u> (13)
- <u>Clean Energy</u> (9)
- <u>Condensed Matter Physics</u> (35)
- Engineering (3)
- Joerg Heber (7)
- Materials Science (31)
- Nanotechnology (34)
- Photonics (33)
- <u>Publishing</u> (7)
- Quantum physics (15)
- <u>Science Communication</u> (5)
- <u>Science Policy</u> (18)
- <u>Thermodynamics</u> (3)

#### Latest tweets

- As Richard Feynman died 25 years ago today, here he is at his most human: I love my wife. My wife is dead. <u>lettersofnote.com/2012/02/i-love...</u> 2 days ago
- Too much suffering... MT @malonebarry: And the entire collection of winning images from World Press Photo is here -> worldpressphoto.org/awards/2013 3 days ago
- Amazing sight of a meteor over Russia this morning RT @gbrumfiel: Holy cow (unconfirmed):

youtube.com/watch?v=iCawTY... 3 days ago

- RT @<u>ngoroff</u>: @joergheber Endgadget has a nuanced view of the Tesla/@<u>nytimes</u> story. engt.co/UkRyuE <u>3 days ago</u>
- RT @jackshafer: If cars are devices designed by geniuses to be used by idiots, the Tesla has a long way to go. <u>3 days ago</u>

Follow @joergheber

#### Blogroll

- <u>2020 Science</u>
- Chemical Connections
- <u>Condensed Concepts</u>
- <u>Cosmic Variance</u>
- Female Science Professor
- Freak of Nature
- nanoscale views
- <u>Nature News Blog</u>
- Not Exactly Rocket Science
- <u>Sean Carroll</u>
- Soft Machines
- The Language of Bad Physics
- <u>The Loom</u>
- The physics arXiv blog
- The Sceptical Chymist
- <u>Uncertain Principles</u>
- Wandering Gaia

#### License



All That Matters is licensed under a Creative Commons Attribution 3.0 Unported License.